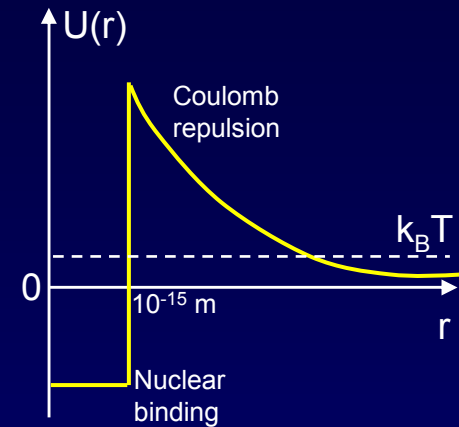


Tunneling Example: The Sun

The solar nuclear fusion process starts when two protons fuse together. In order for this reaction to proceed, the protons must “touch” (approach to within 10^{-15} m of each other).

The potential energy, $U(r)$, looks something like this:



The temperature of the sun’s core is $T \sim 1.3 \times 10^7$ K.

This corresponds to an average kinetic energy:

$$k_B T = 2 \times 10^{-16} \text{ J} \quad (k_B = \text{Boltzman's constant})$$

At $r = 10^{-15}$ m the height of the Coulomb barrier is:

$$U(r) = (1/4\pi\epsilon_0)e^2/r = (9 \times 10^9) \times (1.6 \times 10^{-19} \text{ C})^2 / 10^{-15} \text{ m} \\ = 2 \times 10^{-13} \text{ J}$$

Thus, the protons in the sun very rarely have enough thermal energy to go over the Coulomb barrier.

How do they fuse then? **By tunneling through the barrier!**

