

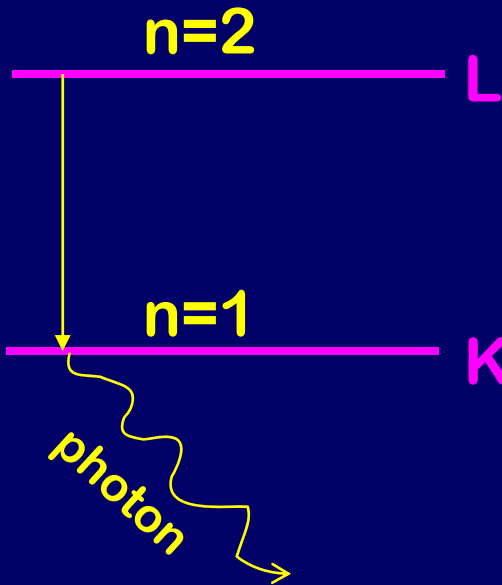
Example

K_{α} X-Rays

Estimate the energy of K_{α} X-rays off of a silver (Ag) target ($Z=47$).

Better formula for multi-electron atoms

$$E_n = \frac{(-13.6)(Z-1)^2}{n^2}$$



$$E_L = -13.6\text{eV}(47-1)^2 \frac{1}{2^2} = -7.2\text{keV}$$

$$E_K = -13.6\text{eV}(47-1)^2 \frac{1}{1^2} = -28.8\text{keV}$$

$$E(K_{\alpha}) = E_L - E_K = 21.6\text{keV}$$

(vs. **21.7 keV** Expt)

Not bad!

