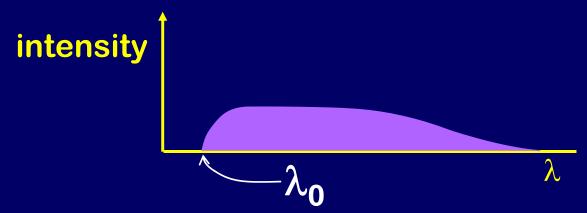
## Bremsstrahlung X-Rays



- Electron hits atom and slows down, losing kinetic energy.
  - Energy emitted as photon
- Electron hitting atom makes many photons (X-Rays), all with different energy.
  - Many different wavelengths.



- If all of electron's energy is lost to a single photon, photon has maximum energy (minimum wavelength).
  - Minimum X-Ray wavelength =  $\lambda_o$ .