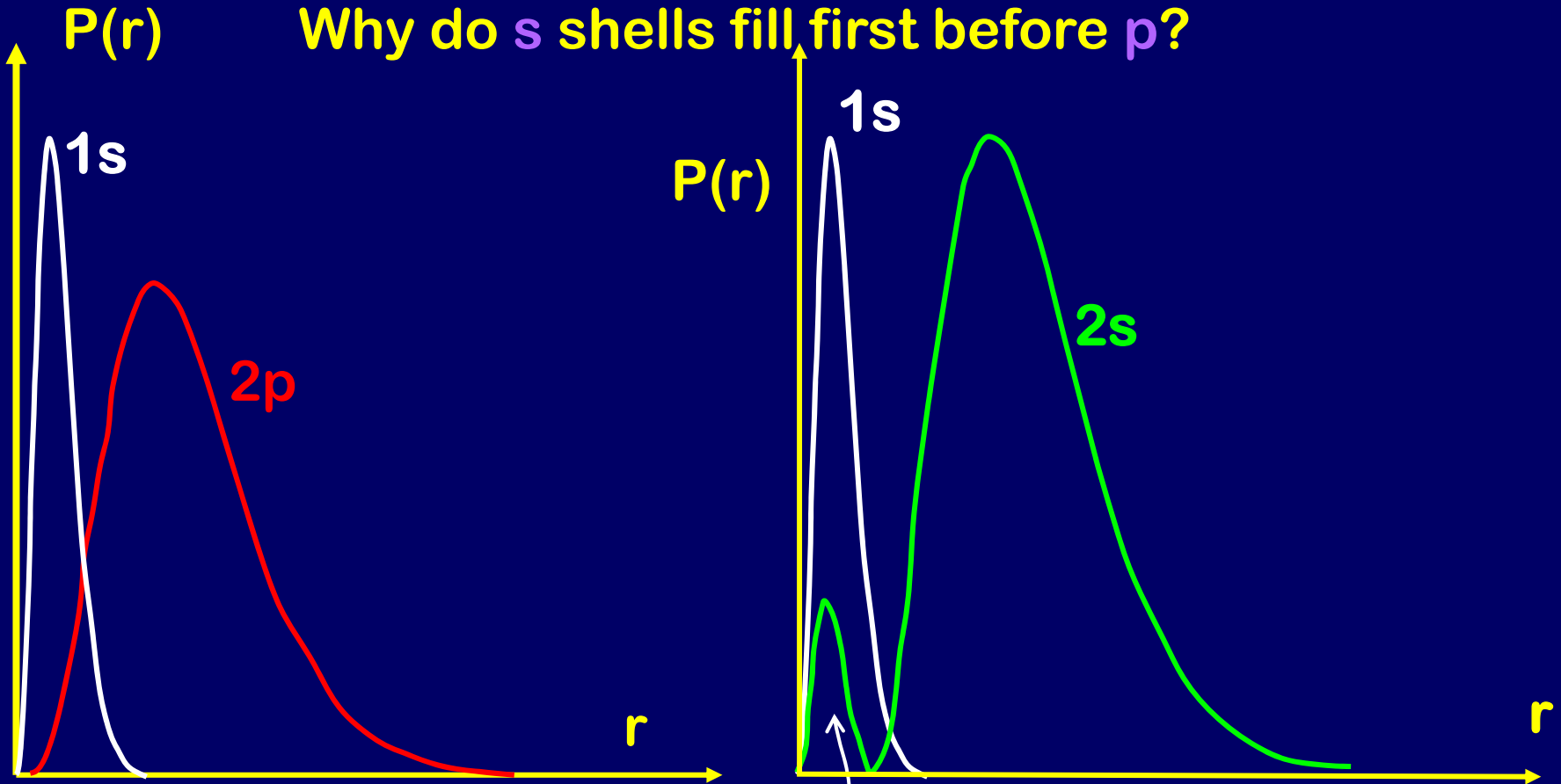


Shell Ordering

Why do s shells fill first before p?



2s electrons can get closer to nucleus, which means less "shielding" from the 1s electrons