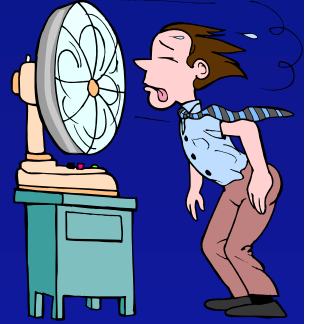
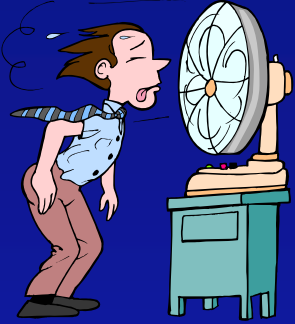


Heat



○ Definition: Flow of energy between two objects due to difference in temperature

□ Note: similar to WORK

□ Object does not “have” heat (it has energy)

○ Units: calorie

□ Amount of heat needed to raise 1g of water 1°C

□ 1 Calorie = 1000 calories = 4186 Joules