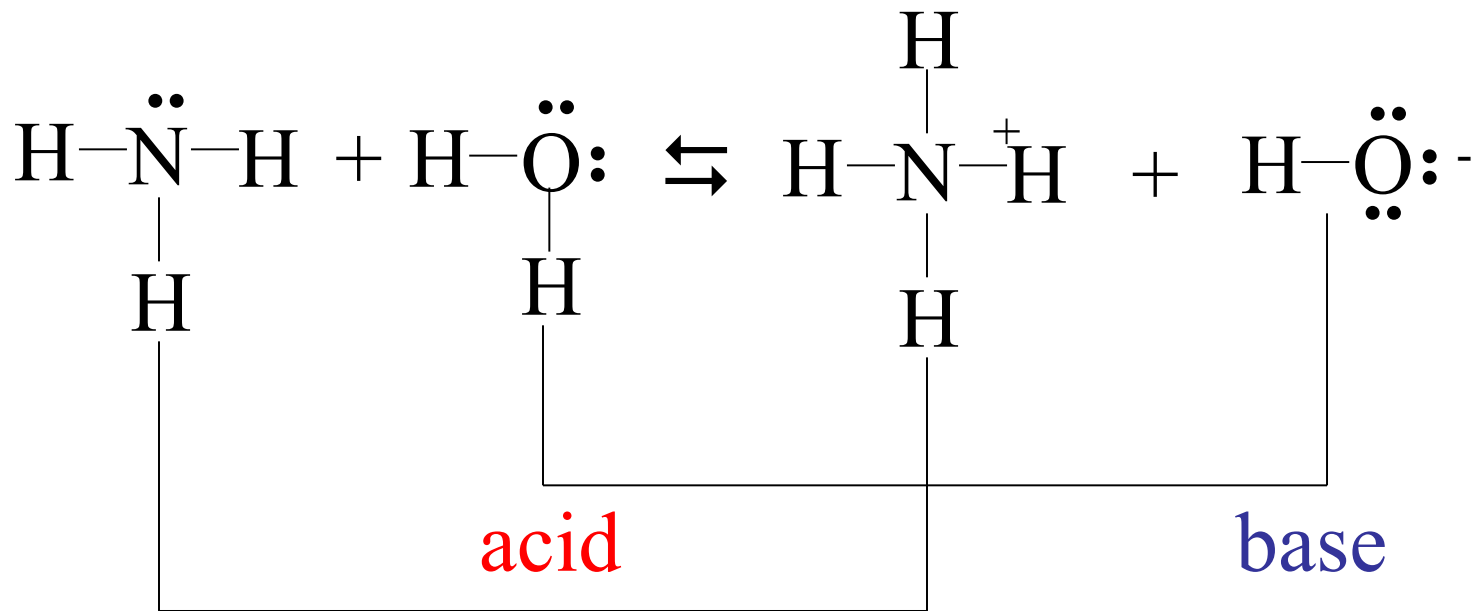


Brønsted-Lowry Acids and Bases

acid proton donor

base proton acceptor



base

acid

weak base

relatively strong **acid**



$$K_b = 1.8 \times 10^{-5}$$