What is the oxidation state of S in each of these compounds:

$$H_{2}S$$
 $S = H = S_{8}$ $S = SCl_{2}$ $S = Cl = SO_{4}^{2-}$ $S = O = SO_{4}^{2-}$

Which of these compounds can **not** act as a **reducing agent**?
Which of these compounds can **not** act as a **oxidizing agent**?