



Determining ΔH_{rxn}

- Since $\Delta H = q_p$, determine the heat transferred at constant pressure. This is termed calorimetry.
- Since energy is conserved, we can often determine ΔH_{rxn} for one reaction given values of ΔH_{rxn} for others. Uses the principle in Hess's Law.
- Use standard enthalpies of formations for reactants and products (we have tables of these).