## Clicker Question

## Given:

$$Na_2CrO_{4(aq)} + AgNO_{3(aq)} \rightarrow$$

Ag<sub>2</sub>CrO<sub>4</sub> + NaNO<sub>3</sub>

Which is the solid?

- A) NaNO<sub>3</sub>
- B) Ag<sub>2</sub>CrO<sub>4</sub>

## Solubility Rules

- Most nitrate salts are soluble.
- Most salts of sodium, potassium, and ammonium cations are soluble.
- Most chloride salts are soluble.
  Exceptions: Ag<sup>+</sup> and Pb<sup>2+</sup>.
- Most sulfate salts are soluble.
  Exceptions: Ca<sup>2+</sup>, Ba<sup>2+</sup>, and Pb<sup>2+</sup>.
- Most hydroxide salts are only slightly soluble. Soluble ones are: Na+, K+, and Ca<sup>2+</sup>.
- Most sulfide, carbonate, and phosphate salts are only slightly soluble.