**Table 5.4** ► Names of Common Polyatomic Ions

lon	Name	lon	Name
NH <sub>4</sub> <sup>+</sup>	ammonium	CO <sub>3</sub> <sup>2-</sup>	carbonate
NO <sub>2</sub> <sup>-</sup>	nitrite	HCO <sub>3</sub>	hydrogen carbonate
NO <sub>3</sub>	nitrate		(bicarbonate is a widely
SO <sub>3</sub> <sup>2-</sup>	sulfite		used common name)
SO <sub>4</sub> <sup>2-</sup>	sulfate	CIO-	hypochlorite
HSO <sub>4</sub>	hydrogen sulfate	CIO <sub>2</sub> <sup>-</sup>	chlorite
11304	(bisulfate is a widely	CIO <sub>3</sub> <sup>-</sup>	chlorate
	used common name)	CIO <sub>4</sub> <sup>-</sup>	perchlorate
OH <sup>-</sup>	hydroxide	$C_2H_3O_2^-$	acetate
CN-	cyanide	MnO <sub>4</sub> <sup>-</sup>	permanganate
PO <sub>4</sub> <sup>3-</sup>	phosphate	Cr <sub>2</sub> O <sub>7</sub> <sup>2-</sup>	dichromate
HPO <sub>4</sub> <sup>2-</sup>	hydrogen phosphate	CrO <sub>4</sub> <sup>2-</sup>	chromate
H <sub>2</sub> PO <sub>4</sub> <sup>-</sup>	dihydrogen phosphate	O <sub>2</sub> <sup>2-</sup>	peroxide

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